

## **Biocidal Products Committee (BPC)**

Opinion on the application for approval of the active substance:

**Silver copper zeolite**

**Product type: 2**

ECHA/BPC/210/2018

Adopted

17 October 2018



## Opinion of the Biocidal Products Committee

### on the application for approval of the active substance Silver copper zeolite for product type 2

In accordance with Article 89(1) of Regulation (EU) No 528/2012 of the European Parliament and of the Council 22 May 2012 concerning the making available on the market and use of biocidal products (BPR), the Biocidal Products Committee (BPC) has adopted this opinion on the non-approval in product type 2 of the following active substance:

<b>Common name:</b>	<b>Silver copper zeolite</b>
<b>Chemical name:</b>	<b>Silver copper zeolite (zeolite, LTA framework type<sup>1</sup>, ion-exchanged with silver, copper and ammonium ions)</b>
<b>EC No.:</b>	<b>not assigned</b>
<b>CAS No.:</b>	<b>130328-19-7<sup>2</sup></b>
<b>Existing active substance</b>	

This document presents the opinion adopted by the BPC, having regard to the conclusions of the evaluating Competent Authority. The assessment report, as a supporting document to the opinion, contains the detailed grounds for the opinion.

### Process for the adoption of BPC opinions

Following the submission of an application by the European Silver Task Force on 17 December 2007, the evaluating Competent Authority Sweden submitted an assessment report and the conclusions of its evaluation to ECHA on 12 June 2017. In order to review the assessment report and the conclusions of the evaluating Competent Authority, the Agency organised consultations via BPC (BPC-27) and its Working Groups (WG V 2017). Revisions agreed upon were presented and the assessment report and the conclusions were amended accordingly.

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<sup>1</sup> The framework type is a crucial part of the identity. A silver copper zeolite with a different framework-type would not be considered the same substance.

<sup>2</sup> The CAS-name is zeolites, synthetic, Ag. The entry in the CAS inventory is broader than the specified chemical name.

## **Adoption of the BPC opinion**

### **Rapporteur: Sweden**

The BPC opinion on the non-approval of the active substance silver copper zeolite in product type 2 was adopted on 17 October 2018.

The BPC opinion was adopted by consensus. The opinion is published on the ECHA webpage at:

<http://echa.europa.eu/regulations/biocidal-products-regulation/approval-of-active-substances/bpc-opinions-on-active-substance-approval>

## Detailed BPC opinion and background

### 1. Overall conclusion

The overall conclusion of the BPC is that the silver copper zeolite in product type 2 may not be approved. The detailed grounds for the overall conclusion are described in the assessment report.

### 2. BPC Opinion

#### 2.1. BPC Conclusions of the evaluation

##### a) Presentation of the active substance including the classification and labelling of the active substance

This evaluation covers the use of silver copper zeolite in product type 2.

Silver copper zeolite (zeolite, LTA framework type, ion-exchanged with silver, copper and ammonium ions) is an inorganic active substance, which cannot be analysed as the complete substance. The specification is thus based on the concentration ranges for major elements as well as maximum levels for elements regarded as impurities. A specification for the reference source is established. Arsenic (As) is regarded as a relevant impurity with a maximum level of 34 ppm.

The physico-chemical properties of the active substance and biocidal product have been evaluated and are deemed acceptable for the intended use, storage and transportation of the active substance and biocidal product.

Validated analytical methods are available for the technical material with respect to major elements as well as the elements regarded as impurities (significant and relevant). Validated analytical monitoring methods for silver and copper are available for the relevant matrices (soil, water and food). The methods for copper (except for food) are from the public domain and have previously been deemed acceptable for the approval of copper carbonate.

A harmonised classification is not available for silver copper zeolite. The Swedish Chemicals Agency has submitted a proposal for harmonised classification and labelling on 3 July 2017.

The proposed classification and labelling for silver copper zeolite according to Regulation (EC) No 1272/2008 (CLP Regulation) is:

<b>Proposed Classification according to the CLP Regulation</b>	
Hazard Class and Category Codes	Repr. 2 Aquatic acute 1 Aquatic chronic 1
<b>Labelling</b>	
Pictogram codes	GHS08 GHS09
Signal Word	warning
Hazard Statement Codes	H361d H400 H410
Specific Concentration limits, M-Factors	M = 100 for acute and chronic

### Justification for the proposal

There is no substance-specific information with respect to fertility effects of silver copper zeolite. In the absence of substance-specific information, a robust classification proposal cannot be presented. However, due to the structural similarity with silver zinc zeolite and the similarity of effects observed with other silver salts that do not contain zinc, it is reasonable to assume that silver copper zeolite meets the criteria for classification Repr. 2; H361d (Suspected of damaging the unborn child), as concluded for silver zinc zeolite in the RAC opinion.

### b) Intended use, target species and effectiveness

Silver copper zeolite is used to treat polymers to achieve an antimicrobial effect. The silver ion is the active species, which is released out of the treated polymer. The silver ion interacts with the cell membrane of microorganisms, interferes with electron transport processes, binds to nucleic acids, inhibits enzymes and catalyses free radical oxygen species.

Generally, the antimicrobial effect of polymer materials containing silver active substances is dependent on how much of the silver is released. A precondition for the release of silver is a solvent, i.e. a liquid which the material comes into contact with. A dry polymer material surface will not release any silver ions and thus will not exert an antimicrobial effect. This is why claims and use-conditions have to be described to be able to demonstrate efficacy. Efficacy has to be demonstrated towards one example use, respectively, for the claims made.

A claim against bacteria has been made. The example uses given were i) wall or floor covering to avoid cross-contamination and ii) air conditioning components to inhibit microbial growth.

For example use i), fast bacteriocidal effects in a dry surrounding would need to be demonstrated. Such tests have not been provided.

For example use ii) bacteriostatic efficacy under wet conditions needs to be demonstrated. However, normally disinfectants for air-conditioning systems are applied by airborne diffusion of an aerosol, a smoke, a vapour or a gas. It would need to be shown with an appropriate test simulating practical conditions of use that the required performance standards can be met by a biocidal product containing Silver copper zeolite incorporated into the parts of an air-conditioning system. Such tests have not been provided. In conclusion, efficacy for this example use is not sufficiently demonstrated.

Efficacy for example use i) or ii) has not been sufficiently demonstrated to recommend approval.

### Resistance

The risk of antibacterial resistance and cross resistance developing from an increased use of silver, in particular new and increasing wide-spread and disperse use in consumer products, cannot be assessed with the currently available information.

## c) Overall conclusion of the evaluation including need for risk management measures

### Human health

For several of the human health endpoints there is no substance-specific data available. However, silver copper zeolite is expected to dissociate during the acidic conditions of the stomach and the constituents of the substance are assumed to be absorbed individually. Therefore, the hazard assessment of silver copper zeolite is made based on data available for each constituent of silver copper zeolite, i.e. silver, copper and the zeolite. The assessment of the silver ion is based on studies in which the silver ion is indirectly tested, i.e. studies performed with silver sodium hydrogen zirconium phosphate, silver chloride and silver acetate. Based on information on silver ion content and release for the silver substances, the dose of silver copper zeolite needed to achieve the same exposure can be calculated. Likewise, the assessment of the copper ion is made based on the assessment report for copper sulfate pentahydrate.

Animal studies indicate a low acute toxicity via oral, dermal and inhalation routes. The substance causes initial and transient skin and eye irritation but effects do not meet criteria for classification. Silver copper zeolite does not cause skin sensitisation reactions in guinea pigs.

In the absence of substance-specific information it is assumed, based on data for silver nitrate, that 5% of the active substance as well as of silver ions released from silver copper zeolite are orally absorbed. Similarly, the dermal absorption is expected to be 5% based on data for silver nitrate.

Effects following subchronic exposure include an increased level of alkaline phosphatase and pigmentation of organs and tissues, effects commonly seen in studies with different silver substances. Results indicate a weak clastogenic potential in vitro but weight of evidence indicates that silver copper zeolite lacks a genotoxic potential in vivo.

There is no substance-specific information on the chronic toxicity and carcinogenic potential of silver copper zeolite. Based on data available for silver zinc zeolite and copper sulfate, the substance is not expected to have a carcinogenic potential.

No developmental toxicity was observed in pups from dams treated with silver copper zeolite up to 2000 mg silver copper zeolite/kg bw/d but there is no substance-specific information on fertility available. Due to the structural similarity with silver zinc zeolite and the similarity of effects observed with other silver salts that do not contain zinc, it is reasonable to assume that silver copper zeolite fulfils criteria for classification Repr. 2; H361d (suspected of damaging the unborn child), as concluded for silver zinc zeolite.

There is no robust information available to assess the neurotoxic or immunotoxic properties of silver copper zeolite or of the read across substances. However, the available data did not show clear indications of such properties.

An assessment of the endocrine disruptor (ED) properties was conducted. However, this ED assessment could not be finalised as the data are considered insufficient for an assessment against the criteria laid down in Regulation (EU) No 2017/2100.

The table below summarises the exposure scenarios assessed.

### **Industrial use**

Scenario	Primary exposure and description of scenarios	Risk acceptable
Mixing and loading	Tier 1	no
	Tier 2 (respiratory protection, 95%)	no
	Tier 2 (protective gloves, 95%)	no
	Tier 2 (respiratory protection, 95%, and protective gloves, 95%)	yes

Mixing and loading without PPE and by using either respiratory protection or protective gloves show unacceptable risks. However, the risk is acceptable for industrial professionals when appropriate personal protection equipments (respiratory protection, 95%, and protective gloves, 95%) are worn.

### **Use in paints and coatings**

Scenario	Primary exposure and description of scenarios	Risk acceptable
Spray application	by professionals with PPE	no
Brush and roll application	by professionals and non-professionals, with PPE	no
Joint sealant application	professionals and non-professionals, without PPE	yes

The risks for professionals and non-professionals when applying paints by spraying, brushing or rolling are not acceptable. PPE equipment is not sufficient to mitigate these risks. However, the risk from primary exposure during joint sealant application is acceptable.

### **Consumer use of biocidal product or solid treated articles<sup>3</sup>**

#### **Non-textile polymers, secondary exposure**

Scenario	Exposure category <sup>4</sup>	Risk acceptable
Articles intended for dermal contact	small-scale, all age-groups	yes
	medium-scale, adults	yes
	medium-scale, infant, toddler, child	no
	large-scale, all age-groups	no
	hand-to mouth contact infant and toddler	yes
Articles intended for oral contact	small-scale adults, children and toddlers	yes
	Large-scale for infants	no
	Large-scale for toddlers, children and adults	yes

<sup>3</sup> Depending on the claim, some of the treated articles might be considered biocidal product.

<sup>4</sup> Large scale, medium scale and small scale exposure categories refer to the duration of dermal exposure and exposed body surface.

Medium-scale use of non-textile polymers with direct skin contact shows unacceptable risks for infants, toddlers and children. Large-scale use shows risks for all age groups. However, medium-scale use is acceptable for adults. Small-scale use is acceptable for all age-groups. Furthermore, hand-to-mouth contact from treated articles is acceptable for infants and toddlers. Large-scale use of polymers intended to be mouthed by infants shows unacceptable risks. However, large-scale use is acceptable for all age-groups as well as small-scale use apart from infants.

### **Textile polymers, secondary exposure**

<b>Scenario</b>	<b>Exposure category</b>	<b>Risk acceptable</b>
Oral exposure to treated textile	Taking into mouth by infants and toddlers	yes
Textiles intended for direct contact with skin	Small-scale, adults, children, infants and toddlers	no
	Large-scale adults, children, infants and toddlers	no
Textiles	Textile handling	yes

Use of textiles intended for direct contact with human skin shows unacceptable risks for all age-groups. However, the mouthing of textiles by infants and toddlers as well as the handling of wet textiles is acceptable.

### **Environment**

Silver copper zeolite releases silver ions ( $\text{Ag}^+$ ) under the use envisaged, which is the active component of silver copper zeolite. Besides silver, also copper ions are released. Thus, environmental fate has been addressed for silver as well as for copper, because both are toxic to environmental organisms. Owing to its use in treated articles, silver copper zeolite does not enter water bodies in its original composition (i.e. silver and copper adsorbed to zeolite). It will dissociate and, thus, the different components silver, copper and zeolite will have different environmental fates. Silver and copper are released from the treated polymers through ion exchange and migration in the presence of aquatic media, whereas the zeolite part is expected to mainly remain in the polymer matrix.

Since copper does not contribute significantly to the environmental toxicity of the active substance, the environmental risk assessment was conducted for silver only.

Emissions to atmosphere are negligible.

No unacceptable risks were identified for sewage treatment plants for the intended uses.

The standard concept of assessing potential for bioaccumulation is not applicable for metals. Trophic transfer can be an important route of exposure, but evidence of significant biomagnification is lacking. No unacceptable risk for secondary poisoning has been identified.

Unacceptable risks for groundwater are not expected for the intended uses.

No further risks for the environment are identified from aggregated exposure to silver copper zeolite, including use in other product types.

**Polymer formulation – industrial use**

Scenario	Aquatic	Terrestrial	Risk acceptable
Polymer formulation (handling, compounding and conversion of polymers from which articles (non-textile polymers and textile polymers) are shaped)	yes	yes	yes

**Biocidal products or solid treated articles<sup>5</sup> - service life**

Scenario	Examples	Aquatic	Terrestrial	Risk acceptable
Non-textile polymers, indoor use	laminated work surface, walls, floors and air conditioning components	yes	yes	yes

**Treated textiles, service life**

Scenario	Examples	Aquatic	Terrestrial	Risk acceptable
Release to environment via laundry	clothing	yes	yes	yes

**Paints, coatings and sealants, application**

Scenario	Aquatic	Terrestrial	Risk acceptable
Sealants indoor	yes	yes	yes

**Paints, coatings and sealants, service life**

Scenario	Aquatic	Terrestrial	Risk acceptable
Sealants indoor	yes	yes	yes

The risk from polymer formulation is acceptable. Use of paint, coatings and sealants indoors and use of treated articles indoors is acceptable.

A substantial contribution of copper to the environmental risk is not expected.

**Overall conclusion**

Sufficient efficacy has not been demonstrated. Thus, approval cannot be suggested.

<sup>5</sup> Depending on the claim, some of the treated articles might be considered biocidal products.

## 2.2. Exclusion, substitution and POP criteria

### 2.2.1. Exclusion and substitution criteria

The table below summarises the relevant information with respect to the assessment of exclusion and substitution criteria:

Property		Conclusions	
CMR properties	Carcinogenicity (C)	no classification required	Silver copper zeolite does not fulfil criterion (a), (b) and (c) of Article 5(1)
	Mutagenicity (M)	no classification required	
	Toxic for reproduction (R)	Repr. Cat. 2	
PBT and vPvB properties	Persistent (P) or very Persistent (vP)	Silver copper zeolite as inorganic metal is excluded from the P assessment taking into account Annex XIII of the REACH Regulation (EU) No 1272/2008.	Silver copper zeolite does not fulfil criterion (e) of Article 5(1) and does not fulfil criterion (d) of Article 10(1)
	Bioaccumulative (B) or very Bioaccumulative (vB)	Silver copper zeolite is not B or vB.	
	Toxic (T)	Silver copper zeolite is T.	
Endocrine disrupting properties	The data available is considered insufficient to assess the endocrine properties of silver copper zeolite. Consequently, no conclusion can be drawn whether silver copper zeolite fulfils criterion (d) of Article 5(1) or criterion (e) of Article 10(1).		
Respiratory sensitisation properties	Silver copper zeolite does not fulfil criterion (b) of Article 10(1). No classification is required		
Concerns linked to critical effects	Silver copper zeolite does not fulfil criterion (e) of Article 10(1).		
Proportion of non-active isomers or impurities	Silver copper zeolite does not fulfil criterion (f) of Article 10(1).		

Consequently, the following is concluded:

Silver copper zeolite does not meet the exclusion criteria laid down in Article 5 of Regulation (EU) No 528/2012.

Silver copper zeolite does not meet the conditions laid down in Article 10 of Regulation (EU) No 528/2012, and is therefore not considered as a candidate for substitution.

The exclusion and substitution criteria were assessed in line with the "Note on the principles for taking decisions on the approval of active substances under the BPR"<sup>6</sup> and in line with "Further guidance on the application of the substitution criteria set out under Article 10(1) of the BPR"<sup>7</sup> agreed at the 54<sup>th</sup> and 58<sup>th</sup> meeting respectively, of the representatives of Member States Competent Authorities for the implementation of Regulation 528/2012 concerning the making available on the market and use of biocidal products. This implies that the assessment of the exclusion criteria is based on Article 5(1) and the assessment of substitution criteria is based on Article 10(1)(a, b, d, e and f). However, the exclusion criteria were not assessed in line with the criteria laid down in the Annex of Regulation (EU) No 2017/2100, which apply as of 7 June 2018.

### **2.2.2. POP criteria**

POP criteria are not applicable for silver copper zeolite, as the substance is inorganic. There are no indications (monitoring data or modelling data) of any long-range transport potential of the active substance either.

### **2.3. BPC opinion on the application for approval of the active substance Silver copper zeolite in product type 2**

In view of the conclusions of the evaluation, it is proposed that silver copper zeolite shall not be approved. The criteria laid down in point (b)(i) of Article 19(1) of Regulation (EU) 528/2012 are not met.

The active substance does not fulfil the criteria according to Article 28(2) to enable inclusion in Annex I of Regulation (EU) 528/2012. Silver copper zeolite gives rise to concern for human health and the environment, i.e. it is proposed to be classified as Repr. 2 and as Aquatic acute 1.

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<sup>6</sup> See document: Note on the principles for taking decisions on the approval of active substances under the BPR (available from <https://circabc.europa.eu/d/a/workspace/SpacesStore/c41b4ad4-356c-4852-9512-62e72cc919df/CA-March14-Doc.4.1%20-%20Final%20-%20Principles%20for%20substance%20approval.doc>)

<sup>7</sup> See document: Further guidance on the application of the substitution criteria set out under article 10(1) of the BPR (available from [https://circabc.europa.eu/d/a/workspace/SpacesStore/dbac71e3-cd70-4ed7-bd40-fc1cb92cfe1c/CA-Nov14-Doc.4.4%20-%20Final%20-%20Further%20guidance%20on%20Art10\(1\).doc](https://circabc.europa.eu/d/a/workspace/SpacesStore/dbac71e3-cd70-4ed7-bd40-fc1cb92cfe1c/CA-Nov14-Doc.4.4%20-%20Final%20-%20Further%20guidance%20on%20Art10(1).doc))